

## Naming Compounds



sodium



chlorine



sodium chloride



# Naming simple compounds

To name simple compounds of metals and nonmetals:

1. Write down the name of the metal.
2. Write down the name of the nonmetal, changing the ending of the word to “-ide”.

What is the name of the compound made when the following elements combine?

- |                        |                        |
|------------------------|------------------------|
| ● magnesium and oxygen | <b>magnesium oxide</b> |
| ● sodium and chlorine  | <b>sodium chloride</b> |
| ● oxygen and iron      | <b>iron oxide</b>      |



# Naming simple compounds

What is the name of each compound formed by these metal and nonmetal elements?

element 1	element 2	compound
iron (Fe)	sulfur (S)	iron sulfide (FeS)
magnesium (Mg)	nitrogen (N)	magnesium nitride (Mg <sub>3</sub> N <sub>2</sub> )
sodium (Na)	chlorine (Cl)	sodium chloride (NaCl)
tin (Sn)	oxygen (O)	tin oxide (SnO)
aluminum (Al)	bromine (Br)	aluminum bromide (AlBr <sub>3</sub> )
nickel (Ni)	iodine (I)	nickel iodide (NiI <sub>2</sub> )
zinc (Zn)	sulfur (S)	zinc sulfide (ZnS)
lithium (Li)	nitrogen (N)	lithium nitride (Li <sub>3</sub> N)

# Naming compounds containing oxygen

Many compounds contain more than two elements.

For compounds containing two elements **plus** oxygen, the ending of the other nonmetal usually changes to “**-ate**”.

element 1	element 2	element 3	compound
nickel	sulfur	oxygen	nickel sulfate
magnesium	nitrogen	oxygen	magnesium nitrate
sodium	nitrogen	oxygen	sodium nitrate
copper	sulfur	oxygen	copper sulfate
aluminum	bromine	oxygen	aluminum bromate



What is the name of the compound produced?

① sodium + nitrogen + oxygen →

?

② aluminum + bromine →

?

③ lithium + nitrogen →

?

④ aluminum + bromine + oxygen →

?

aluminum bromate

?

C

solve

↶

# Writing a word equation

A word equation can be used to describe any chemical reaction.

The steps for writing a word equation are:

1. On the left-hand side, put the name(s) of the **reactant(s)**. If there are two or more reactants, link them with a **+** sign.
2. In the middle, draw an **arrow**: **→**
3. On the right-hand side, put the name(s) of the **product(s)**. If there are two or more products, link them with a **+** sign.

**reactant 1 + reactant 2 → product 1 + product 2**



Fill in the gaps in these word equations

